

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (\AA) for bonds involving the elements H, B, C, N, O, F, Si, P, S, Cl, As, Se, Br, Te, and I

Bond	Substructure	d	m	σ	q_l	q_u	n	Note
As(3)—As(3)	$X_2-\text{As}-\text{As}-X_2$	2.459	2.457	0.011	2.456	2.466	8	
As—B	see CUDLOC (2.065), CUDLUI (2.041)							
As—Br	see CODDEE, CODDII (2.346–3.203)							
As(4)—C	$X_3-\text{As}-\text{CH}_3$ $(X)_2(\text{C}, \text{O}, \text{S}=\text{O})\text{As}-\text{C}_{sp^3}$ $\text{As}-\text{C}_{ar}$ in Ph_4As^+ $(X)_2(\text{C}, \text{O}, \text{S}=\text{O})\text{As}-\text{C}_{ar}$	1.903	1.907	0.016	1.893	1.916	12	
		1.927	1.929	0.017	1.921	1.937	16	
		1.905	1.909	0.012	1.897	1.912	108	
		1.922	1.927	0.016	1.908	1.934	36	
As(3)—C	$X_2-\text{As}-\text{C}_{sp^3}$ $X_2-\text{As}-\text{C}_{ar}$	1.963	1.965	0.017	1.948	1.978	6	
		1.956	1.956	0.015	1.944	1.964	41	
As(3)—Cl	$X_2-\text{As}-\text{Cl}$	2.268	2.256	0.039	2.247	2.281	10	
As(6)—F	in AsF_6^-	1.678	1.676	0.020	1.659	1.695	36	
As(3)—I	see OPIMAS (2.579, 2.590)							
As(3)—N(3)	$X_2-\text{As}-\text{N}-X_2$	1.858	1.858	0.029	1.839	1.873	19	
As(4)=N(2)	see TPASSN (1.837)							
As(4)—O	$(X)_2(\text{O}=\text{O})\text{As}-\text{OH}$	1.710	1.712	0.017	1.695	1.726	6	
As(3)—O	see ASAZOC, PHASOC01 (1.787–1.845)							
As(4)=O	$X_3-\text{As}=\text{O}$	1.661	1.661	0.016	1.652	1.667	9	
As(3)—P(3)	see BELNIP (2.350, 2.362)							†
As(3)=P(3)	see BUTHAZ10 (2.124)							†
As(3)—S	$X_2-\text{As}-\text{S}$	2.275	2.266	0.032	2.247	2.298	14	
As(4)=S	$X_3-\text{As}=\text{S}$	2.083	2.082	0.004	2.080	2.086	9	
As(3)—Se(2)	see COSDIX, ESEARS (2.355–2.401)							†
As(3)—Si(4)	see BICGEZ, MESIAD (2.351–2.365)							†
As(3)—Te(2)	see ETEARS (2.571, 2.576)							†
B(n)—B(n)	$n = 5-7$ in boron cages	1.775	1.773	0.031	1.763	1.786	688	
B(4)—B(4)	see CETTAW (2.041)							
B(4)—B(3)	see COFVOI (1.698)							
B(3)—B(3)	$X_2-\text{B}-\text{B}-X_2$	1.701	1.700	0.014	1.691	1.712	8	
B(6)—Br		1.967	1.971	0.014	1.954	1.979	7	†
B(4)—Br		2.017	2.008	0.031	1.990	2.044	15	†
B(n)—C	$n = 5-7$: $\text{B}-\text{C}$ in cages $n = 3-4$: $\text{B}-\text{C}_{sp^3}$ not cages $n = 4$: $\text{B}-\text{C}_{ar}$ B^--C_{ar} in Ph_4B^- $n = 3$: $\text{B}-\text{C}_{ar}$	1.716	1.717	0.020	1.707	1.728	96	I
		1.597	1.599	0.022	1.585	1.611	29	
		1.606	1.607	0.012	1.596	1.615	41	
		1.643	1.643	0.006	1.641	1.645	16	
		1.556	1.552	0.015	1.546	1.566	24	
B(n)—Cl	$\text{B}(5)-\text{Cl}$ and $\text{B}(3)-\text{Cl}$ $\text{B}(4)-\text{Cl}$	1.751	1.751	0.011	1.743	1.761	14	
		1.833	1.833	0.013	1.821	1.843	22	
B(4)—F	$\text{B}-\text{F}$ (B neutral) B^--F in BF_4^-	1.366	1.368	0.017	1.356	1.375	25	
		1.365	1.372	0.029	1.352	1.390	84	

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
B(4)—I	see TMPBTI (2.220, 2.253)							
B(4)—N(3)	$X_3-\mathbf{B}-\mathbf{N}(=\text{C})(X)$ in pyrazaboles	1.611 1.549	1.617 1.552	0.013 0.015	1.601 1.536	1.625 1.560	8 10	
B(3)—N(3)	$X_2-\mathbf{B}-\mathbf{N}-\text{C}_2$: all coplanar for $\tau(\text{BN}) > 30^\circ$ see BOGSUL, BUSHAY, CILRUK (1.434–1.530) $\text{S}_2-\mathbf{B}-\mathbf{N}-X_2$	1.404 1.447	1.404 1.443	0.014 0.013	1.389 1.435	1.408 1.470	40 14	2
B(4)—O	$\mathbf{B}-\mathbf{O}$ in BO_4^- for neutral $\mathbf{B}-\mathbf{O}$ see Note 3	1.468	1.468	0.022	1.453	1.479	24	3
B(3)—O(2)	$X_2-\mathbf{B}-\mathbf{O}-X$	1.367	1.367	0.024	1.349	1.382	35	
B(<i>n</i>)—P	<i>n</i> = 4: $\mathbf{B}-\mathbf{P}$ <i>n</i> = 3: see BUPSIB10 (1.892, 1.893)	1.922	1.927	0.027	1.900	1.954	10	
B(4)—S	$\mathbf{B}(4)-\text{S}(3)$ $\mathbf{B}(4)-\text{S}(2)$	1.930 1.896	1.927 1.896	0.009 0.004	1.925 1.893	1.934 1.899	10 6	
B(3)—S	$\text{N}-\mathbf{B}-\text{S}_2$ $(=\text{X}-)(\text{N}-)\mathbf{B}-\text{S}$	1.806 1.851	1.806 1.854	0.010 0.013	1.799 1.842	1.816 1.859	28 10	
Br—Br	see BEPZEB, TPASTB	2.542	2.548	0.015	2.526	2.551	4	
Br—C	$\mathbf{Br}-\text{C}^*$ $\mathbf{Br}-\text{C}sp^3$ (cyclopropane) $\mathbf{Br}-\text{C}sp^2$ $\mathbf{Br}-\text{C}_{ar}$ (mono-Br + <i>m,p</i> -Br ₂) $\mathbf{Br}-\text{C}_{ar}$ (<i>o</i> -Br ₂)	1.966 1.910 1.883 1.899 1.875	1.967 1.910 1.881 1.899 1.872	0.029 0.010 0.015 0.012 0.011	1.951 1.900 1.874 1.892 1.864	1.983 1.914 1.894 1.906 1.884	100 8 31 119 8	4 8 4 4 4
⁻ Br(2)—Cl	see TEACBR (2.362–2.402)							†
Br—I	see DTHIBR10 (2.646), TPHOSI (2.695)							
Br—N	see NBBZAM (1.843)							
Br—O	see CIYFOF	1.581	1.581	0.007	1.574	1.587	4	
Br—P	see CISTED (2.366)							
Br—S(2)	see BEMLIO (2.206)							†
Br—S(3)	see CIWYIQ (2.435, 2.453)							†
Br—S(3) ⁺	see THINBR (2.321)							†
Br—Se	see CIFZUM (2.508, 2.619)							
Br—Si	see BIZJAV (2.284)							
Br—Te	In $\text{Br}_6\text{Te}^{2-}$ see CUGBAH (2.692–2.716) $\mathbf{Br}-\text{Te}(4)$ see BETUTE10 (3.079, 3.015) $\mathbf{Br}-\text{Te}(3)$ see BTUPTE (2.835)							
$\text{C}sp^3-\text{C}sp^3$	$\text{C}^\#-\text{CH}_2-\text{CH}_3$ $(\text{C}^\#)_2-\text{CH}-\text{CH}_3$ $(\text{C}^\#)_3-\text{C}-\text{CH}_3$ $\text{C}^\#-\text{CH}_2-\text{CH}_2-\text{C}^\#$ $(\text{C}^\#)_2-\text{CH}-\text{CH}_2-\text{C}^\#$ $(\text{C}^\#)_3-\text{C}-\text{CH}_2-\text{C}^\#$ $(\text{C}^\#)_2-\text{CH}-\text{CH}-(\text{C}^\#)_2$ $(\text{C}^\#)_3-\text{C}-\text{CH}-(\text{C}^\#)_2$ $(\text{C}^\#)_3-\text{C}-\text{C}-(\text{C}^\#)_3$ C^*-C^* (overall)	1.513 1.524 1.534 1.524 1.531 1.538 1.542 1.556 1.588 1.530	1.514 1.526 1.534 1.524 1.531 1.539 1.542 1.556 1.580 1.530	0.014 0.015 0.011 0.014 0.012 0.010 0.011 0.011 0.025 0.015	1.507 1.518 1.527 1.516 1.524 1.533 1.536 1.549 1.566 1.521	1.523 1.534 1.541 1.532 1.538 1.544 1.549 1.562 1.610 1.539	192 226 825 2459 1217 330 321 215 21 5777	5, 6

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	q_l	q_u	<i>n</i>	Note
$Csp^3 - Csp^3$ (cont.)	in cyclopropane (any substituent)	1.510	1.509	0.026	1.497	1.523	888	7
	in cyclobutane (any substituent)	1.554	1.553	0.021	1.540	1.567	679	8
	in cyclopentane (C,H substituents)	1.543	1.543	0.018	1.532	1.554	1641	
	in cyclohexane (C,H substituents)	1.535	1.535	0.016	1.525	1.545	2814	
	cyclopropyl—C* (exocyclic)	1.518	1.518	0.019	1.505	1.531	366	7
	cyclobutyl—C* (exocyclic)	1.529	1.529	0.016	1.519	1.539	376	8
	cyclopentyl—C* (exocyclic)	1.540	1.541	0.017	1.527	1.549	956	
	cyclohexyl—C* (exocyclic)	1.539	1.538	0.016	1.529	1.549	2682	
	in cyclobutene (any substituent)	1.573	1.574	0.017	1.566	1.586	25	8
	in cyclopentene (C,H substituents)	1.541	1.539	0.015	1.532	1.549	208	
	in cyclohexene (C,H substituents)	1.541	1.541	0.020	1.528	1.554	586	
	in oxirane (epoxide)	1.466	1.466	0.015	1.458	1.474	249	9
	in aziridine	1.480	1.481	0.021	1.465	1.496	67	9
	in oxetane	1.541	1.541	0.019	1.527	1.557	16	
	in azetidine	1.548	1.543	0.018	1.536	1.558	22	
oxiranyl—C* (exocyclic)	1.509	1.507	0.018	1.497	1.519	333	9	
aziridinyl—C* (exocyclic)	1.512	1.512	0.018	1.496	1.526	13	9	
$Csp^3 - Csp^2$	$CH_3 - C=C$	1.503	1.504	0.011	1.497	1.509	215	
	$C^\# - CH_2 - C=C$	1.502	1.502	0.013	1.494	1.510	483	
	$(C^\#)_2 - CH - C=C$	1.510	1.510	0.014	1.501	1.518	564	
	$(C^\#)_3 - C - C=C$	1.522	1.522	0.016	1.511	1.533	193	
	$C^* - C=C$ (overall)	1.507	1.507	0.015	1.499	1.517	1456	5
	$C^* - C=C$ (endocyclic):							
	in cyclopropene	1.509	1.508	0.016	1.500	1.516	20	10
	in cyclobutene	1.513	1.512	0.018	1.500	1.525	50	8
	in cyclopentene	1.512	1.512	0.014	1.502	1.521	208	
	in cyclohexene	1.506	1.505	0.016	1.495	1.516	391	
	in cyclopentadiene	1.502	1.503	0.019	1.490	1.515	18	
	in cyclohexa-1,3-diene	1.504	1.504	0.017	1.491	1.517	56	
	$C^* - C\equiv C$ (exocyclic):							
	cyclopropenyl—C*	1.478	1.475	0.012	1.470	1.485	7	10
	cyclobutenyl—C*	1.489	1.483	0.015	1.479	1.496	11	8
	cyclopentenyl—C*	1.504	1.506	0.012	1.495	1.512	115	
	cyclohexenyl—C*	1.511	1.511	0.013	1.502	1.519	292	
	$C^* - CH=O$ in aldehydes	1.510	1.510	0.008	1.501	1.518	7	
	$(C^*)_2 - C=O$ in ketones	1.511	1.511	0.015	1.501	1.521	952	11
	in cyclobutanone	1.529	1.530	0.016	1.514	1.545	18	
	in cyclopentanone	1.514	1.514	0.016	1.505	1.523	312	
acyclic and 6+ rings	1.509	1.509	0.016	1.499	1.519	626		
$C^* - COOH$ in carboxylic acids	1.502	1.502	0.014	1.495	1.510	176		
$C^* - COO^-$ in carboxylate anions	1.520	1.521	0.011	1.516	1.528	57		
$C^* - C(=O)(-OC^*)$ in acyclic esters	1.497	1.496	0.018	1.484	1.509	553	12	
in β -lactones	1.519	1.519	0.020	1.500	1.538	4	13	
in γ -lactones	1.512	1.512	0.015	1.501	1.521	110	12	
in δ -lactones	1.504	1.502	0.013	1.495	1.517	27	12	
cyclopropyl (C)—C=O in ketones, acids, and esters	1.486	1.485	0.018	1.474	1.497	105	7	
$C^* - C(=O)(-NH_2)$ in acyclic amides	1.514	1.512	0.016	1.506	1.526	32	14	
$C^* - C(=O)(-NHC^*)$ in acyclic amides	1.506	1.505	0.012	1.498	1.515	78	14	
$C^* - C(=O)[-N(C^*)_2]$ in acyclic amides	1.505	1.505	0.011	1.496	1.517	15	14	
$Csp^3 - C_{ar}$	$CH_3 - C_{ar}$	1.506	1.507	0.011	1.501	1.513	454	
	$C^\# - CH_2 - C_{ar}$	1.510	1.510	0.009	1.505	1.516	674	
	$(C^\#)_2 - C - C_{ar}$	1.515	1.515	0.011	1.508	1.522	363	
	$(C^\#)_3 - C - C_{ar}$	1.527	1.530	0.016	1.517	1.539	308	
	$C^* - C_{ar}$ (overall)	1.513	1.513	0.014	1.505	1.521	1813	
	cyclopropyl (C)—C _{ar}	1.490	1.490	0.015	1.479	1.503	90	7

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note	
<i>Csp</i> ³ — <i>Csp</i> ¹	C*—C≡C	1.466	1.465	0.010	1.460	1.469	21	15	
	C [#] —C≡C	1.472	1.472	0.012	1.464	1.481	88	15	
	C*—C≡N	1.470	1.469	0.013	1.463	1.479	106	7(b)	
	cyclopropyl (C)—C≡N	1.444	1.447	0.010	1.436	1.451	38	7	
<i>Csp</i> ² — <i>Csp</i> ²	C=C—C=C (conjugated)	1.455	1.455	0.011	1.447	1.463	30	16, 18	
	(unconjugated)	1.478	1.476	0.012	1.470	1.479	8	17, 18	
	(overall)	1.460	1.460	0.015	1.450	1.470	38		
	C=C—C=C—C=C	1.443	1.445	0.013	1.431	1.454	29	18	
	C=C—C=C (endocyclic in TCNQ)	1.432	1.433	0.012	1.424	1.441	280	19	
	C=C—C(=O)(—C*) (conjugated)	1.464	1.462	0.018	1.453	1.476	211	16, 18	
	(unconjugated)	1.484	1.486	0.017	1.475	1.497	14	17, 18	
	(overall)	1.465	1.462	0.018	1.453	1.478	226		
	C=C—C(=O)—C=C:								
	in benzoquinone (C,H substituents only)	1.478	1.476	0.011	1.469	1.488	28		
	in benzoquinone (any substituent)	1.478	1.478	0.031	1.464	1.498	172		
	non-quinonoid	1.456	1.455	0.012	1.447	1.464	28		
	C—C—COOH	1.475	1.476	0.015	1.461	1.488	22		
	C=C—COOC*	1.488	1.489	0.014	1.478	1.497	113		
	C=C—COO ⁻	1.502	1.499	0.017	1.488	1.510	11		
	HOOC—COOH	1.538	1.537	0.007	1.535	1.541	9		
HOOC—COO ⁻	1.549	1.552	0.009	1.546	1.553	13			
⁻ OOC—COO ⁻	1.564	1.559	0.022	1.554	1.568	9			
formal <i>Csp</i> ² — <i>Csp</i> ² single bond in selected, non-fused heterocycles:									
in 1 <i>H</i> -pyrrole (C3—C4)	1.412	1.410	0.016	1.401	1.427	29			
in furan (C3—C4)	1.423	1.423	0.016	1.412	1.433	62			
in thiophene (C3—C4)	1.424	1.425	0.015	1.415	1.433	40			
in pyrazole (C3—C4)	1.410	1.412	0.016	1.400	1.418	20			
in isoxazole (C3—C4)	1.425	1.425	0.016	1.413	1.438	9			
in furazan (C3—C4)	1.428	1.427	0.007	1.422	1.435	6			
in furoxan (C3—C4)	1.417	1.417	0.006	1.412	1.422	14			
<i>Csp</i> ² — <i>C_{ar}</i>	C=C—C _{ar} (conjugated)	1.470	1.470	0.015	1.463	1.480	37	16, 18	
	(unconjugated)	1.488	1.490	0.012	1.480	1.496	87	17, 18	
	(overall)	1.483	1.483	0.015	1.472	1.494	124		
	cyclopropenyl (C=C)—C _{ar}	1.447	1.448	0.006	1.441	1.452	8	10	
	C _{ar} —C(=O)—C*	1.488	1.489	0.016	1.478	1.500	84		
	C _{ar} —C(=O)—C _{ar}	1.480	1.481	0.017	1.468	1.494	58		
	C _{ar} —COOH	1.484	1.485	0.014	1.474	1.491	75		
	C _{ar} —C(=O)(—OC*)	1.487	1.487	0.012	1.480	1.494	218		
	C _{ar} —COO ⁻	1.504	1.509	0.014	1.495	1.512	26		
	C _{ar} —C(=O)—NH ₂	1.500	1.503	0.020	1.498	1.510	19		
	C _{ar} —C=N—C [#] (conjugated)	1.476	1.478	0.014	1.466	1.486	27	16	
	(unconjugated)	1.491	1.490	0.008	1.485	1.496	48	17	
	(overall)	1.485	1.487	0.013	1.481	1.493	75		
in indole (C3—C3a)	1.434	1.434	0.011	1.428	1.439	40			
<i>Csp</i> ² — <i>Csp</i> ¹	C=C—C≡C	1.431	1.427	0.014	1.425	1.441	11	7(b)	
	C=C—C≡N in TCNQ	1.427	1.427	0.010	1.420	1.433	280	19	
<i>C_{ar}</i> — <i>C_{ar}</i>	in biphenyls (<i>ortho</i> substituent all H)	1.487	1.488	0.007	1.484	1.493	30		
	(≥ 1 non-H <i>ortho</i> substituent)	1.490	1.491	0.010	1.486	1.495	212		
<i>C_{ar}</i> — <i>Csp</i> ¹	C _{ar} —C≡C	1.434	1.436	0.006	1.430	1.437	37		
	C _{ar} —C≡N	1.443	1.444	0.008	1.436	1.448	31		
<i>Csp</i> ¹ — <i>Csp</i> ¹	C≡C—C≡C	1.377	1.378	0.012	1.374	1.384	21		
<i>Csp</i> ² = <i>Csp</i> ²	C*—CH=CH ₂	1.299	1.300	0.027	1.280	1.311	42		
	(C*) ₂ —C=CH ₂	1.321	1.321	0.013	1.313	1.328	77		
	C*—CH=CH—C* (<i>cis</i>)	1.317	1.318	0.013	1.310	1.323	106		
	(<i>trans</i>)	1.312	1.311	0.011	1.304	1.320	19		
	(overall)	1.316	1.317	0.015	1.309	1.323	127		

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	q_l	q_u	<i>n</i>	Note	
$Csp^2=Csp^2$ (cont.)	$(C^*)_2-C=CH-C^*$	1.326	1.328	0.011	1.319	1.334	168		
	$(C^*)_2-C=C-(C^*)_2$	1.331	1.330	0.009	1.326	1.334	89		
	$(C^*,H)_2-C=C-(C^*,H)_2$ (overall)	1.322	1.323	0.014	1.315	1.331	493	5	
	in cyclopropene (any substituent)	1.294	1.288	0.017	1.284	1.302	10	10	
	in cyclobutene (any substituent)	1.335	1.335	0.019	1.324	1.347	25	8	
	in cyclopentene (C,H substituents)	1.323	1.324	0.013	1.314	1.331	104		
	in cyclohexene (C,H substituents)	1.326	1.325	0.012	1.318	1.334	196		
	$C=C=C$ (allenes, any substituents)	1.307	1.307	0.005	1.303	1.310	18		
	$C=C-C=C$ (C,H substituents, conjugated)	1.330	1.330	0.014	1.322	1.338	76	16	
	$C=C-C=C-C=C$ (C,H substituents, conjugated)	1.345	1.345	0.012	1.337	1.350	58	16	
	$C=C-C_{ar}$ (C,H substituents, conjugated)	1.339	1.340	0.011	1.334	1.346	124	16	
	$C=C$ in cyclopenta-1,3-diene (any substituent)	1.341	1.341	0.017	1.328	1.356	18		
	$C=C$ in cyclohexa-1,3-diene (any substituent)	1.332	1.332	0.013	1.323	1.341	56		
	in $C=C-C=O$ (C,H substituent, conjugated)	1.340	1.340	0.013	1.332	1.348	211	16, 18	
(C,H substituent, unconjugated)	1.331	1.330	0.008	1.326	1.339	14	17, 18		
(C,H substituent, overall)	1.340	1.339	0.013	1.332	1.348	226			
in cyclohexa-2,5-dien-1-ones	1.329	1.327	0.011	1.321	1.335	28			
in <i>p</i> -benzoquinones (C*,H substituents)	1.333	1.337	0.011	1.325	1.338	14			
(any substituent)	1.349	1.339	0.030	1.330	1.364	86			
in TCNQ (endocyclic)	1.352	1.353	0.010	1.345	1.358	142	19		
(exocyclic)	1.392	1.391	0.017	1.379	1.405	139	19		
$C=C-OH$ in enol tautomers	1.362	1.360	0.020	1.349	1.370	54			
in heterocycles (any substituent)									
1 <i>H</i> -pyrrole (C2—C3, C4—C5)	1.375	1.377	0.018	1.361	1.388	58			
furan (C2—C3, C4—C5)	1.341	1.342	0.021	1.329	1.351	125			
thiophene (C2—C3, C4—C5)	1.362	1.359	0.025	1.346	1.377	60			
pyrazole (C4—C5)	1.369	1.372	0.019	1.362	1.383	20			
imidazole (C4—C5)	1.360	1.361	0.014	1.352	1.367	44			
isoxazole (C4—C5)	1.341	1.336	0.012	1.331	1.355	9			
indole (C2—C3)	1.364	1.363	0.012	1.355	1.371	40			
$C_{ar} \equiv C_{ar}$	in phenyl rings with C*,H substituents only:								
	H—C≡C—H	1.380	1.381	0.013	1.372	1.388	2191		
	C*—C≡C—H	1.387	1.388	0.010	1.382	1.393	891		
	C*—C≡C—C*	1.397	1.397	0.009	1.392	1.403	182		
	C≡C (overall)	1.384	1.384	0.013	1.375	1.391	3264		
	F—C≡C—F	1.372	1.374	0.011	1.366	1.380	84	4	
	Cl—C≡C—Cl	1.388	1.389	0.014	1.380	1.398	152	4	
	in naphthalene (D_{2h}) C1—C2	1.364	1.364	0.014	1.356	1.373	440		
	(any substituent) C2—C3	1.406	1.406	0.014	1.397	1.415	218		
	C1—C8a	1.420	1.419	0.012	1.412	1.426	440		
	C4a—C8a	1.422	1.424	0.011	1.417	1.429	109		
	in anthracene (D_{2h}) C1—C2	1.356	1.356	0.009	1.350	1.360	56		
	(any substituent) C2—C3	1.410	1.410	0.010	1.401	1.416	34		
	C1—C9a	1.430	1.430	0.006	1.426	1.434	56		
	C4a—C9a	1.435	1.436	0.007	1.429	1.440	34		
	C9—C9a	1.400	1.402	0.009	1.395	1.406	68		
	in pyridine (C,H substituent)	1.379	1.381	0.012	1.371	1.387	276	20	
	(any substituents)	1.380	1.380	0.015	1.371	1.389	537	20	
	in pyridinium cation:								
	(N ⁺ —H; C,H substituents on C) C2—C3	1.373	1.375	0.012	1.368	1.380	30		
	C3—C4	1.379	1.380	0.011	1.371	1.388	30		
(N ⁺ —X; C,H substituents on C) C2—C3	1.373	1.372	0.019	1.362	1.382	151			
C3—C4	1.383	1.385	0.019	1.372	1.394	151			
in pyrazine (H substituent on C)	1.379	1.377	0.010	1.370	1.388	10			
(any substituent on C)	1.405	1.405	0.024	1.388	1.420	60			
in pyrimidine (C,H substituents on C)	1.387	1.389	0.018	1.379	1.400	28			

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note	
<i>Csp</i> ¹ ≡ <i>Csp</i> ¹	<i>X</i> —C≡C— <i>X</i>	1.183	1.183	0.014	1.174	1.193	119	15	
	C,H—C≡C—C,H	1.181	1.181	0.014	1.173	1.192	104	15	
	in C≡C—C(<i>sp</i> ² , ar)	1.189	1.193	0.010	1.181	1.195	38	15	
	in C≡C—C≡C	1.192	1.192	0.010	1.187	1.197	42	15	
	in CH≡C—C [#]	1.174	1.174	0.011	1.167	1.180	42	15	
<i>Csp</i> ³ —Cl	Omitting 1,2-dichlorides: C—CH ₂ —Cl	1.790	1.790	0.007	1.783	1.795	13	4	
	C ₂ —CH—Cl	1.803	1.802	0.003	1.800	1.807	8	4	
	C ₃ —C—Cl	1.849	1.856	0.011	1.837	1.858	5	4	
	<i>X</i> —CH ₂ —Cl (<i>X</i> = C,H,N,O)	1.790	1.791	0.011	1.783	1.797	37	4	
	<i>X</i> ₂ —CH—Cl (<i>X</i> = C,H,N,O)	1.805	1.803	0.014	1.800	1.812	26	4	
	<i>X</i> ₃ —C—Cl (<i>X</i> = C,H,N,O)	1.843	1.838	0.014	1.835	1.858	7	4	
	<i>X</i> ₂ —C—Cl ₂ (<i>X</i> = C,H,N,O)	1.779	1.776	0.015	1.769	1.790	18	4	
	<i>X</i> —C—Cl ₃ (<i>X</i> = C,H,N,O)	1.768	1.765	0.011	1.761	1.776	33	4	
	Cl—CH(—C)—CH(—C)—Cl	1.793	1.793	0.013	1.786	1.800	66	4	
	Cl—C(—C ₂)—C(—C ₂)—Cl	1.762	1.760	0.010	1.757	1.765	54	4	
	cyclopropyl—Cl	1.755	1.756	0.011	1.749	1.763	64	4	
	<i>Csp</i> ² —Cl	C=C—Cl (C,H,N,O substituents on C)	1.734	1.729	0.019	1.719	1.748	63	4
		C=C—Cl ₂ (C,H,N,O substituents on C)	1.720	1.716	0.013	1.708	1.729	20	4
		Cl—C=C—Cl	1.713	1.711	0.011	1.705	1.720	80	4
<i>C</i> _{ar} —Cl	<i>C</i> _{ar} —Cl (mono-Cl + <i>m,p</i> -Cl ₂)	1.739	1.741	0.010	1.734	1.745	340	4	
	<i>C</i> _{ar} —Cl (<i>o</i> -Cl ₂)	1.720	1.720	0.010	1.713	1.717	364	4	
<i>Csp</i> ¹ —Cl	see HCLENE10 (1.634, 1.646)								
<i>Csp</i> ³ —F	Omitting 1,2-difluorides: C—CH ₂ —F and C ₂ —CH—F	1.399	1.399	0.017	1.389	1.408	25	4	
	C ₃ —C—F	1.428	1.431	0.009	1.421	1.435	11	4	
	(C*,H) ₂ —C—F ₂	1.349	1.347	0.012	1.342	1.356	58	4	
	C*—C—F ₃	1.336	1.334	0.007	1.330	1.344	12	4	
	F—C*—C*—F	1.371	1.374	0.007	1.362	1.375	26	4	
	<i>X</i> ₃ —C—F (<i>X</i> = C,H,N,O)	1.386	1.389	0.033	1.373	1.408	70	4	
	<i>X</i> ₂ —C—F ₂ (<i>X</i> = C,H,N,O)	1.351	1.349	0.013	1.342	1.356	58	4	
	<i>X</i> —C—F ₃ (<i>X</i> = C,H,N,O)	1.322	1.323	0.015	1.314	1.332	309	4	
	F—C(— <i>X</i>) ₂ —C(— <i>X</i>) ₂ —F (<i>X</i> = C,H,N,O)	1.373	1.374	0.009	1.362	1.377	30	4	
	F—C(— <i>X</i>) ₂ —NO ₂ (<i>X</i> = any substituent)	1.320	1.319	0.009	1.312	1.327	18	4	
<i>Csp</i> ² —F	C=C—F (C,H,N,O substituents on C)	1.340	1.340	0.013	1.334	1.346	34	4	
<i>C</i> _{ar} —F	<i>C</i> _{ar} —F (mono-F + <i>m,p</i> -F ₂)	1.363	1.362	0.008	1.357	1.368	38	4	
	<i>C</i> _{ar} —F (<i>o</i> -F ₂)	1.340	1.340	0.009	1.336	1.344	167	4	
<i>Csp</i> ³ —H	C—C—H ₃ (methyl)	1.059	1.061	0.030	1.039	1.083	83	21	
	C ₂ —C—H ₂ (primary)	1.092	1.095	0.013	1.088	1.099	100	21	
	C ₃ —C—H (secondary)	1.099	1.097	0.004	1.095	1.103	14	21	
	C _{2,3} —C—H (primary and secondary)	1.093	1.095	0.012	1.089	1.100	118	21	
	<i>X</i> —C—H ₃ (methyl)	1.066	1.074	0.028	1.049	1.087	160	21	
	<i>X</i> ₂ —C—H ₂ (primary)	1.092	1.095	0.012	1.088	1.099	230	21	
	<i>X</i> ₃ —C—H (secondary)	1.099	1.099	0.007	1.095	1.103	117	21	
	<i>X</i> _{2,3} —C—H (primary and secondary)	1.094	1.096	0.011	1.091	1.100	348	21	
<i>Csp</i> ² —H	C—C=C—H	1.077	1.079	0.012	1.074	1.085	14	21	
<i>C</i> _{ar} —H	<i>C</i> _{ar} —H	1.083	1.083	0.011	1.080	1.087	218	21	
<i>Csp</i> ³ —I	C*—I	2.162	2.159	0.015	2.149	2.179	15	4	
<i>C</i> _{ar} —I	<i>C</i> _{ar} —I	2.095	2.095	0.015	2.089	2.104	51	4	
<i>Csp</i> ³ —N(4)	C*—NH ₃ ⁺	1.488	1.488	0.013	1.482	1.495	298		
	(C*) ₂ —NH ₂ ⁺	1.494	1.493	0.016	1.484	1.503	249		
	(C*) ₃ —NH ⁺	1.502	1.502	0.015	1.491	1.512	509		
	(C*) ₄ —N ⁺	1.510	1.509	0.020	1.496	1.523	319		
	C*—N ⁺ (overall)	1.499	1.498	0.018	1.488	1.510	1370		

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	q_l	q_u	<i>n</i>	Note	
$Csp^3-N(3)$	C^*-N^+ in N-substituted pyridinium	1.485	1.484	0.009	1.477	1.490	32		
	C^*-NH_2 (Nsp^3 : pyramidal)	1.469	1.470	0.010	1.462	1.474	19	22	
	$(C^*)_2-NH$ (Nsp^3 : pyramidal)	1.469	1.467	0.012	1.461	1.477	152	5, 22	
	$(C^*)_3-N$ (Nsp^3 : pyramidal)	1.469	1.468	0.014	1.460	1.476	1042	5, 22	
	C^*-Nsp^3 (overall)	1.469	1.468	0.014	1.460	1.476	1201		
	Csp^3-Nsp^3 in aziridine in azetidione in tetrahydropyrrole in piperidine		1.472	1.471	0.016	1.464	1.482	134	
			1.484	1.481	0.018	1.472	1.495	21	
			1.475	1.473	0.016	1.464	1.483	66	
			1.473	1.473	0.013	1.460	1.479	240	
	Csp^3-Nsp^2 (N planar) in: acyclic amides $C^*-NH-C=O$ β -lactams $C^*-N(-X)-C=O$ (<i>endo</i>) γ -lactams $C^*-NH-C=O$ (<i>endo</i>) $C^*-N(-C^*)-C=O$ (<i>endo</i>) $C^*-N(-C^*)-C=O$ (<i>exo</i>) δ -lactams $C^*-NH-C=O$ (<i>endo</i>) $C^*-N(-C^*)-C=O$ (<i>endo</i>) $C^*-N(-C^*)-C=O$ (<i>exo</i>) nitro compounds (1,2-dinitro omitted): $C-CH_2-NO_2$ $C_2-CH-NO_2$ C_3-C-NO_2 $C_2-C-(NO_2)_2$ 1,2-dinitro: $NO_2-C^*-C^*-NO_2$		1.454	1.451	0.011	1.446	1.461	78	23 14
			1.464	1.465	0.012	1.458	1.475	23	13
			1.457	1.458	0.011	1.449	1.465	20	13
			1.462	1.461	0.010	1.453	1.466	15	13
			1.458	1.456	0.014	1.448	1.465	15	13
			1.478	1.472	0.016	1.467	1.491	6	14
			1.479	1.476	0.007	1.475	1.482	15	14
			1.468	1.471	0.009	1.462	1.477	15	14
			1.485	1.483	0.020	1.478	1.502	8	
			1.509	1.509	0.011	1.502	1.511	12	
			1.533	1.533	0.013	1.530	1.539	17	
		1.537	1.536	0.016	1.525	1.550	19		
		1.552	1.550	0.023	1.536	1.572	32		
$Csp^3-N(2)$		$C^{\#}-N=N$	1.493	1.493	0.020	1.477	1.506	54	
	$C^{\#}-N=C-C_{ar}$	1.465	1.468	0.011	1.461	1.472	75		
$Csp^2-N(3)$	$C=C-NH_2$ Nsp^2 planar	1.336	1.344	0.017	1.317	1.348	10	23	
	$C=C-NH-C^{\#}$ Nsp^2 planar	1.339	1.340	0.016	1.327	1.351	17	23	
	$C=C-N-(C^{\#})_2$ Nsp^2 planar	1.355	1.358	0.014	1.341	1.363	22	23	
	Nsp^3 pyramidal	1.416	1.418	0.018	1.397	1.432	18	22	
	Csp^2-Nsp^2 (N planar) in: acyclic amides $NH_2-C=O$ acyclic amides $C^*-NH-C=O$ acyclic amides $(C^*)_2-N-C=O$ β -lactams $C^*-NH-C=O$ γ -lactams $C^*-NH-C=O$ γ -lactams $C^*-N(-C^*)-C=O$ δ -lactams $C^*-NH-C=O$ δ -lactams $C^*-N(-C^*)-C=O$ peptides $C^{\#}-N(-X)-C(-C^{\#})(=O)$ ureas $(NH_2)_2-C=O$ ureas $(C^{\#}-NH)_2-C=O$ ureas $[C^{\#}-N]_2-C=O$ thioureas $(X_2N)_2-C=S$ imides $[C^{\#}-C(=O)]_2-NH$ $[C^{\#}-C(=O)]_2-N-C^{\#}$ $[Csp^2-C(=O)]_2-N-C^{\#}$ $[Csp^2-C(=O)]_2-N-Csp^2$ guanidinium $[C-(NH_2)_3]^+$ (unsubstituted) (any substituent) in heterocyclic systems (any substituent): 1 <i>H</i> -pyrrole (N1-C2, N1-C5) indole (N1-C2) pyrazole (N1-C5) imidazole (N1-C2) imidazole (N1-C5)		1.325	1.323	0.009	1.318	1.331	32	23 14
			1.334	1.333	0.011	1.326	1.343	78	14
			1.346	1.342	0.011	1.339	1.356	5	14
			1.385	1.388	0.019	1.374	1.396	23	13
			1.331	1.331	0.011	1.326	1.337	20	13
			1.347	1.344	0.014	1.335	1.359	15	13
			1.334	1.334	0.006	1.330	1.339	6	14
			1.352	1.353	0.010	1.344	1.356	15	14
			1.333	1.334	0.013	1.326	1.340	380	24
			1.334	1.334	0.008	1.329	1.339	48	25, 26
			1.347	1.345	0.010	1.341	1.354	26	25
			1.363	1.359	0.014	1.354	1.370	40	25, 27
			1.346	1.343	0.023	1.328	1.361	192	
			1.376	1.377	0.012	1.369	1.383	64	
			1.389	1.383	0.017	1.376	1.404	38	
			1.396	1.396	0.010	1.389	1.403	46	
		1.409	1.406	0.020	1.391	1.419	28		
		1.321	1.320	0.008	1.314	1.327	39		
		1.328	1.325	0.015	1.317	1.333	140		
			1.372	1.374	0.016	1.363	1.384	58	
		1.370	1.370	0.012	1.364	1.377	40		
		1.357	1.359	0.012	1.347	1.365	20		
		1.349	1.349	0.018	1.338	1.358	44		
		1.370	1.370	0.010	1.365	1.377	44		
$Csp^2-N(2)$	in imidazole (N3-C4)	1.376	1.377	0.011	1.369	1.384	44		
$C_{ar}-N(4)$	$C_{ar}-N^+(C,H)_3$	1.465	1.466	0.007	1.461	1.470	23		

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
C _{ar} —N(3)	C _{ar} —NH ₂ (Nsp ² : planar)	1.355	1.360	0.020	1.340	1.372	33	23
	(Nsp ³ : pyramidal)	1.394	1.396	0.011	1.385	1.403	25	22
	(overall)	1.375	1.377	0.025	1.363	1.394	98	28
	C _{ar} —NH—C [#] (Nsp ² : planar)	1.353	1.353	0.007	1.347	1.359	16	23
	(Nsp ³ : pyramidal)	1.419	1.423	0.017	1.412	1.432	8	22
	(overall)	1.380	1.364	0.032	1.353	1.412	31	28
	C _{ar} —N—(C [#]) ₂ (Nsp ² : planar)	1.371	1.370	0.016	1.363	1.382	41	23
	(Nsp ³ : pyramidal)	1.426	1.425	0.011	1.421	1.431	22	22
	(overall)	1.390	1.385	0.030	1.366	1.420	69	28
	in indole (N1—C7a)	1.372	1.372	0.007	1.367	1.376	40	
C _{ar} —NO ₂	1.468	1.469	0.014	1.460	1.476	556		
C _{ar} —N(2)	C _{ar} —N=N	1.431	1.435	0.020	1.422	1.442	26	
Csp ² =N(3)	in furoxan (⁺ N2=C3)	1.316	1.316	0.009	1.311	1.324	14	
Csp ² =N(2)	C _{ar} —C=N—C [#]	1.279	1.279	0.008	1.275	1.285	75	
	(C,H) ₂ —C=N—OH in oximes	1.281	1.280	0.013	1.273	1.288	67	
	S—C=N—X	1.302	1.302	0.021	1.285	1.319	36	
	in pyrazole (N2=C3)	1.329	1.331	0.014	1.315	1.339	20	
	in imidazole (C2=N3)	1.313	1.314	0.011	1.307	1.319	44	
	in isoxazole (N2=C3)	1.314	1.315	0.009	1.305	1.320	9	
	in furazan (N2=C3, C4=N5)	1.298	1.299	0.006	1.294	1.303	12	
	in furoxan (C4=N5)	1.304	1.306	0.008	1.300	1.308	14	
C _{ar} ≡N(3)	C≡N ⁺ —H (pyrimidinium)	1.335	1.334	0.015	1.325	1.342	30	
	C≡N ⁺ —C* (pyrimidinium)	1.346	1.346	0.010	1.340	1.352	64	
	C≡N ⁺ —O ⁻ (pyrimidinium)	1.362	1.359	0.013	1.353	1.369	56	
C _{ar} ≡N(2)	C≡N (pyridine)	1.337	1.338	0.012	1.300	1.344	269	
	C≡N (pyrazine)	1.336	1.335	0.022	1.319	1.347	120	
	C≡N≡C (pyrimidine)	1.339	1.338	0.015	1.333	1.342	28	
	N≡C≡N (pyrimidine)	1.333	1.335	0.013	1.326	1.337	28	
	C≡N (pyrimidine) (overall)	1.336	1.337	0.014	1.331	1.339	56	
	in any six-membered N-containing aromatic ring:							
	H—C≡N≡C—H	1.334	1.334	0.014	1.327	1.341	146	
	H—C≡N≡C—C*	1.339	1.341	0.013	1.336	1.345	38	
	C*—C≡N≡C—C*	1.345	1.345	0.008	1.342	1.348	24	
	C≡N≡C (overall)	1.336	1.337	0.014	1.329	1.344	204	
Csp ¹ ≡N(2)	X—N ⁺ ≡C ⁻ (isocyanide)	1.144	1.147	0.006	1.140	1.148	6	
Csp ¹ ≡N(1)	C*—C≡N	1.136	1.137	0.010	1.131	1.142	140	19
	C=C—C≡N in TCNQ	1.144	1.144	0.008	1.139	1.149	284	
	C _{ar} —C≡N	1.138	1.138	0.007	1.133	1.143	31	
	X—S—C≡N	1.144	1.141	0.012	1.138	1.151	10	
	(S—C≡N) ⁻	1.155	1.156	0.012	1.147	1.165	14	
Csp ³ —O(2)	in alcohols:							
	CH ₃ —OH	1.413	1.414	0.018	1.395	1.425	17	
	C—CH ₂ —OH	1.426	1.426	0.011	1.420	1.431	75	
	C ₂ —CH—OH	1.432	1.431	0.011	1.425	1.439	266	
	C ₃ —C—OH	1.440	1.440	0.012	1.432	1.449	106	
	C*—OH (overall)	1.432	1.431	0.013	1.424	1.441	464	
	in dialkyl ethers:							
	CH ₃ —O—C*	1.416	1.418	0.016	1.405	1.426	110	29
	C—CH ₂ —O—C*	1.426	1.424	0.011	1.418	1.435	34	
	C ₂ —CH—O—C*	1.429	1.430	0.010	1.420	1.437	53	
	C ₃ —C—O—C*	1.452	1.450	0.011	1.445	1.458	39	
	C*—O—C* (overall)	1.426	1.425	0.019	1.414	1.437	236	
	in aryl alkyl ethers:							
	CH ₃ —O—C _{ar}	1.424	1.424	0.012	1.417	1.431	616	29
	C—CH ₂ —O—C _{ar}	1.431	1.430	0.013	1.422	1.438	188	

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
<i>Csp²</i> —O(2) (cont.)	in aryl alkyl ethers (cont.) C ₂ —CH—O—C _{ar} C ₃ —C—O—C _{ar} C*—O—C _{ar} (overall)	1.447 1.470 1.429	1.446 1.469 1.427	0.020 0.018 0.018	1.435 1.456 1.419	1.466 1.483 1.436	58 55 917	
	in alkyl esters of carboxylic acids: CH ₃ —O—C(=O)—C* C—CH ₂ —O—C(=O)—C* C ₂ —CH—O—C(=O)—C* C ₃ —C—O—C(=O)—C* C*—O—C(=O)—C* (overall)	1.448 1.452 1.460 1.477 1.450	1.449 1.453 1.460 1.475 1.451	0.010 0.009 0.010 0.008 0.014	1.442 1.445 1.454 1.472 1.442	1.455 1.458 1.465 1.484 1.459	200 32 78 6 314	12, 29
	in alkyl esters of α , β -unsaturated acids: C*—O—C(=O)—C=C (overall)	1.453	1.452	0.013	1.444	1.459	112	
	in alkyl esters of benzoic acid C*—O—C(=O)—C(phenyl) (overall)	1.454	1.454	0.012	1.446	1.463	219	
	in ring systems: oxirane (epoxide) (any substituent) oxetane (any substituent) tetrahydrofuran (C,H substituents) tetrahydropyran (C,H substituents) β -lactones: C*—O—C(=O) γ -lactones: C*—O—C(=O) δ -lactones: C*—O—C(=O)	1.446 1.463 1.442 1.441 1.492 1.464 1.461	1.446 1.460 1.441 1.442 1.494 1.464 1.464	0.014 0.015 0.017 0.015 0.010 0.012 0.017	1.438 1.451 1.430 1.431 1.481 1.455 1.452	1.456 1.474 1.451 1.451 1.501 1.473 1.473	498 16 154 22 4 110 27	9 16 12 12
	O—C—O systems in <i>gem</i> -diols, and pyranose and furanose sugars: HO—C*—OH	1.397	1.401	0.012	1.388	1.405	18	30, 31
	C ₅ —O ₅ —C ₁ —O ₁ H in pyranoses: O ₁ axial (α): C ₅ —O ₅ O ₅ —C ₁ C ₁ —O ₁ O ₁ equatorial (β): C ₅ —O ₅ O ₅ —C ₁ C ₁ —O ₁ $\alpha + \beta$ (overall): C ₅ —O ₅ O ₅ —C ₁ C ₁ —O ₁	1.439 1.427 1.403 1.435 1.430 1.393 1.439 1.430 1.401	1.440 1.426 1.400 1.436 1.431 1.393 1.440 1.429 1.399	0.008 0.012 0.012 0.008 0.010 0.007 0.008 0.012 0.011	1.432 1.421 1.391 1.429 1.424 1.386 1.432 1.421 1.392	1.445 1.432 1.412 1.440 1.436 1.399 1.446 1.436 1.407	29 29 29 17 17 17 60 60 60	
	C ₄ —O ₄ —C ₁ —O ₁ H in furanoses: (overall values) C ₄ —O ₄ (overall values) O ₄ —C ₁ (overall values) C ₁ —O ₁	1.442 1.432 1.404	1.446 1.432 1.405	0.012 0.012 0.013	1.436 1.421 1.397	1.449 1.443 1.409	18 18 18	
	C ₅ —O ₅ —C ₁ —O ₁ —C* in pyranoses: O ₁ axial (α): C ₅ —O ₅ O ₅ —C ₁ C ₁ —O ₁ O ₁ —C* O ₁ equatorial (β): C ₅ —O ₅ O ₅ —C ₁ C ₁ —O ₁ O ₁ —C* $\alpha + \beta$ (overall): C ₅ —O ₅ O ₅ —C ₁ C ₁ —O ₁ O ₁ —C*	1.439 1.417 1.409 1.435 1.434 1.424 1.390 1.437 1.436 1.419 1.402 1.436	1.438 1.417 1.409 1.435 1.435 1.424 1.390 1.438 1.436 1.419 1.403 1.436	0.010 0.009 0.014 0.013 0.006 0.008 0.011 0.013 0.009 0.011 0.016 0.013	1.433 1.410 1.401 1.427 1.429 1.418 1.381 1.428 1.431 1.412 1.391 1.428	1.446 1.424 1.417 1.443 1.439 1.431 1.400 1.445 1.442 1.426 1.413 1.445	67 67 67 67 39 39 39 39 126 126 126 126	
	C ₄ —O ₄ —C ₁ —O ₁ —C* in furanoses: (overall values) C ₄ —O ₄ (overall values) O ₄ —C ₁ (overall values) C ₁ —O ₁ (overall values) O ₁ —C*	1.443 1.421 1.410 1.439	1.445 1.418 1.409 1.437	0.013 0.012 0.014 0.014	1.429 1.413 1.401 1.429	1.453 1.431 1.420 1.449	23 23 23 23	
	Miscellaneous: C#—O—SiX ₃ C*—O—SO ₂ —C	1.416 1.465	1.416 1.461	0.017 0.014	1.405 1.454	1.428 1.475	29 33	

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	q_l	q_u	<i>n</i>	Note	
$C_{sp^2}-O(2)$ (cont.)	in enols: $C=C-OH$	1.333	1.331	0.017	1.324	1.342	53		
	in enol esters: $C=C-O-C^*$	1.354	1.353	0.016	1.341	1.363	40		
	in acids: $C^*-C(=O)-OH$	1.308	1.311	0.019	1.298	1.320	174		
	$C=C-C(=O)-OH$	1.293	1.295	0.019	1.279	1.307	22		
	$C_{ar}-C(=O)-OH$	1.305	1.311	0.020	1.291	1.317	75		
	in esters: $C^*-C(=O)-O-C^*$	1.336	1.337	0.014	1.328	1.346	551	12, 29	
	$C=C-C(=O)-O-C^*$	1.332	1.331	0.011	1.324	1.339	112		
	$C_{ar}-C(=O)-O-C^*$	1.337	1.335	0.013	1.329	1.344	219	12	
	$C^*-C(=O)-O-C=C$	1.362	1.359	0.018	1.351	1.374	26		
	$C^*-C(=O)-O-C=C$	1.407	1.405	0.017	1.394	1.420	26		
	$C^*-C(=O)-O-C_{ar}$	1.360	1.359	0.011	1.355	1.367	40	12	
	in anhydrides: $O=C-O-C=O$	1.386	1.386	0.011	1.379	1.393	70		
	in ring systems:								
	furan ($O1-C2, O1-C5$)	1.368	1.369	0.015	1.359	1.377	125		
	isoxazole ($O1-C5$)	1.354	1.354	0.010	1.345	1.360	9		
	β -lactones: $C^*-C(=O)-O-C^*$	1.359	1.359	0.013	1.348	1.371	4	13	
γ -lactones: $C^*-C(=O)-O-C^*$	1.350	1.349	0.012	1.342	1.359	110	12		
δ -lactones: $C^*-C(=O)-O-C^*$	1.339	1.339	0.016	1.332	1.347	27	12		
$C_{ar}-O(2)$	in phenols: $C_{ar}-OH$	1.362	1.364	0.015	1.353	1.373	511		
	in aryl alkyl ethers: $C_{ar}-O-C^*$	1.370	1.370	0.011	1.363	1.377	920	29, 32	
	in diaryl ethers: $C_{ar}-O-C_{ar}$	1.384	1.381	0.014	1.375	1.391	132		
	in esters: $C_{ar}-O-C(=O)-C^*$	1.401	1.401	0.010	1.394	1.408	40	12	
$C_{sp^2}=O(1)$	in aldehydes and ketones:								
	$C^*-CH=O$	1.192	1.912	0.005	1.188	1.197	7		
	$(C^*)_2-C=O$	1.210	1.210	0.008	1.206	1.215	474	5	
	$(C^#)_2-C=O$ in cyclobutanones	1.198	1.198	0.007	1.194	1.204	12		
	in cyclopentanones	1.208	1.208	0.007	1.203	1.212	155		
	in cyclohexanones	1.211	1.211	0.009	1.207	1.216	312		
	$C=C-C=O$	1.222	1.222	0.010	1.216	1.229	225		
	$(C=C)_2-C=O$	1.233	1.229	0.010	1.226	1.242	28		
	$C_{ar}-C=O$	1.221	1.218	0.014	1.212	1.229	85		
	$(C_{ar})_2-C=O$	1.230	1.226	0.015	1.220	1.238	66		
	$C=O$ in benzoquinones	1.222	1.220	0.013	1.211	1.231	86		
	delocalized double bonds in carboxylate anions:								
	$H-C \cdots O_2^-$ (formate)	1.242	1.243	0.012	1.234	1.252	24		
	$C^*-C \cdots O_2^-$	1.254	1.253	0.010	1.247	1.261	114		
	$C=C-C \cdots O_2^-$	1.250	1.248	0.017	1.238	1.261	52		
	$C_{ar}-C \cdots O_2^-$	1.255	1.253	0.010	1.249	1.262	22		
	$HOOC-C \cdots O_2^-$ (hydrogen oxalate)	1.243	1.247	0.015	1.232	1.256	26		
	$-O_2 \cdots C-C \cdots O_2^-$ (oxalate)	1.251	1.251	0.007	1.248	1.254	18		
	in carboxylic acids ($X-COOH$):								
	$C^*-C(=O)-OH$	1.214	1.214	0.019	1.203	1.224	175		
	$C=C-C(=O)-OH$	1.229	1.226	0.017	1.218	1.237	22		
	$C_{ar}-C(=O)-OH$	1.226	1.223	0.020	1.211	1.241	75		
	in esters:								
	$C^*-C(=O)-O-C^*$	1.196	1.196	0.010	1.190	1.202	551	12	
	$C=C-C(=O)-O-C^*$	1.199	1.198	0.009	1.193	1.203	113		
	$C_{ar}-C(=O)-O-C^*$	1.202	1.201	0.009	1.196	1.207	218	12	
	$C^*-C(=O)-O-C=C$	1.190	1.190	0.014	1.184	1.198	26		
	$C^*-C(=O)-O-C_{ar}$	1.187	1.188	0.011	1.181	1.195	40	12	
	in anhydrides: $O=C-O-C=O$	1.187	1.187	0.010	1.184	1.193	70		
	in β -lactones: $C^*-C(=O)-O-C^*$	1.193	1.193	0.006	1.187	1.198	4	13	
	γ -lactones: $C^*-C(=O)-O-C^*$	1.201	1.202	0.009	1.196	1.206	109	12	
	δ -lactones: $C^*-C(=O)-O-C^*$	1.205	1.207	0.008	1.201	1.209	27	12	
	in amides:								
$NH_2-C(-C^*)=O$	1.234	1.233	0.012	1.225	1.243	32	14		
$(C^*-)(C^*,H-)N-C(-C^*)=O$	1.231	1.231	0.012	1.224	1.238	378	14		
β -lactams: $C^*-NH-C=O$	1.198	1.200	0.012	1.193	1.204	23	13		

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	q_l	q_u	<i>n</i>	Note
$Csp^2=O(1)$ (cont.)	in amides (cont.)							
	γ -lactams: $C^*-NH-C=O$	1.235	1.235	0.008	1.232	1.240	20	13
	γ -lactams: $C^*-N(-C^*)-C=O$	1.225	1.226	0.011	1.217	1.233	15	13
	δ -lactams: $C^*-NH-C=O$	1.240	1.241	0.003	1.237	1.243	6	14
	δ -lactams: $O^*-N(-C^*)-C=O$	1.233	1.233	0.007	1.229	1.239	15	14
	in ureas:							
$(NH_2)_2-C=O$	1.256	1.256	0.007	1.249	1.261	24	25, 26	
$(C^{\#}-NH)_2-C=O$	1.241	1.237	0.011	1.235	1.245	13	25	
$[(C^{\#})_n-N]_2-C=O$	1.230	1.230	0.007	1.224	1.234	20	25, 27	
$Csp^3-P(4)$	$C_3-P^+-C^*$	1.800	1.802	0.015	1.790	1.812	35	33
	$C_2-P(=O)-CH_3$	1.791	1.790	0.006	1.786	1.795	10	
	$C_2-P(=O)-CH_2-C$	1.806	1.806	0.009	1.801	1.813	45	
	$C_2-P(=O)-CH-C_2$	1.821	1.821	0.009	1.815	1.828	15	
	$C_2-P(=O)-C-C_3$	1.841	1.842	0.008	1.835	1.847	14	
	$C_2-P(=O)-C^*$ (overall)	1.813	1.811	0.017	1.800	1.822	84	
$Csp^3-P(3)$	C_2-P-C^*	1.855	1.857	0.019	1.840	1.870	23	
$C_{ar}-P(4)$	$C_3-P^+-C_{ar}$	1.793	1.792	0.011	1.786	1.800	276	
	$C_2-P(=O)-C_{ar}$	1.801	1.802	0.011	1.796	1.807	98	
	$Ph_3-P=N^+=P-Ph_3$	1.795	1.795	0.008	1.789	1.800	197	
$C_{ar}-P(3)$	C_2-P-C_{ar}	1.836	1.837	0.010	1.830	1.844	102	
	$(N\cdots)_2P-C_{ar}$ ($P\cdots N$ aromatic)	1.795	1.793	0.011	1.788	1.803	43	
$Csp^3-S(4)$	C^*-SO_2-C ($C^* = CH_3$ excluded)	1.786	1.782	0.018	1.774	1.797	75	
	C^*-SO_2-C (overall)	1.779	1.778	0.020	1.764	1.790	94	
	C^*-SO_2-O-X	1.745	1.744	0.009	1.738	1.754	7	34
	$C^*-SO_2-N-X_2$	1.758	1.736	0.018	1.746	1.773	17	34
$Csp^3-S(3)$	$C^*-S(=O)-C$ ($C^* = CH_3$ excluded)	1.818	1.814	0.024	1.802	1.829	69	
	$C^*-S(=O)-C$ (overall)	1.809	1.806	0.025	1.793	1.820	88	
	$CH_3-S^+-X_2$	1.786	1.787	0.007	1.779	1.792	21	
	$C^*-S^+-X_2$ ($C^* = CH_3$ excluded)	1.823	1.820	0.016	1.812	1.834	18	
	$C^*-S^+-C_2$ (overall)	1.804	1.794	0.025	1.788	1.820	41	
$Csp^3-S(2)$	C^*-SH	1.808	1.805	0.010	1.800	1.819	6	
	CH_3-S-C^*	1.789	1.787	0.008	1.784	1.794	9	
	$C-CH_2-S-C^*$	1.817	1.816	0.013	1.808	1.824	92	
	$C_2-CH-S-C^*$	1.819	1.819	0.011	1.811	1.825	32	
	$C_3-C-S-C^*$	1.856	1.860	0.011	1.854	1.863	26	
	C^*-S-C^* (overall)	1.819	1.817	0.019	1.809	1.827	242	
	in thiirane	1.834	1.835	0.025	1.810	1.858	4	9
	in thietane: see ZCMXSP (1.817, 1.844)							
	in tetrahydrothiophene	1.827	1.826	0.018	1.811	1.837	20	
	in tetrahydrothiopyran	1.823	1.821	0.014	1.812	1.832	24	
$C-CH_2-S-S-X$	1.823	1.820	0.014	1.813	1.832	41		
$C_3-C-S-S-X$	1.863	1.865	0.015	1.848	1.878	11		
$C^*-S-S-X$ (overall)	1.833	1.828	0.022	1.818	1.848	59		
$Csp^2-S(2)$	$C=C-S-C^*$	1.751	1.755	0.017	1.740	1.764	61	
	$C=C-S-C=C$ (in tetrathiafulvalene)	1.741	1.741	0.011	1.733	1.750	88	
	$C=C-S-C=C$ (in thiophene)	1.712	1.712	0.013	1.703	1.722	60	
	$O=C-S-C^{\#}$	1.762	1.759	0.018	1.747	1.778	20	
$C_{ar}-S(4)$	$C_{ar}-SO_2-C$	1.763	1.764	0.009	1.756	1.769	96	
	$C_{ar}-SO_2-O-X$	1.752	1.750	0.008	1.749	1.756	27	
	$C_{ar}-SO_2-N-X_2$	1.758	1.759	0.013	1.749	1.765	106	35
$C_{ar}-S(3)$	$C_{ar}-S(=O)-C$	1.790	1.790	0.010	1.783	1.798	41	
	$C_{ar}-S^+-X_2$	1.778	1.779	0.010	1.771	1.787	10	
$C_{ar}-S(2)$	$C_{ar}-S-C^*$	1.773	1.774	0.009	1.765	1.779	44	
	$C_{ar}-S-C_{ar}$	1.768	1.767	0.010	1.762	1.774	158	

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
C _{ar} —S(2) (cont.)	C _{ar} —S—C _{ar} (in phenothiazine) C _{ar} —S—S—X	1.764	1.764	0.008	1.760	1.769	48	
		1.777	1.777	0.012	1.767	1.785	47	
Csp ¹ —S(2)	N≡C—S—X	1.679	1.683	0.026	1.645	1.698	10	
Csp ¹ —S(1)	(N≡C—S) ⁻	1.630	1.630	0.014	1.619	1.641	14	
Csp ² =S(1)	(C*) ₂ —C=S: see IPMUDS (1.599)							
	(C _{ar}) ₂ —C=S: see CELDOM (1.611)							
	(X) ₂ —C=S (X = C,N,O,S)	1.671	1.675	0.024	1.656	1.689	245	
	X ₂ N—C(=S)—S—X	1.660	1.660	0.016	1.648	1.674	38	
	(X ₂ N) ₂ —C=S (thioureas)	1.681	1.684	0.020	1.669	1.693	96	
	N—C(⋯S) ₂	1.720	1.721	0.012	1.709	1.731	20	
Csp ³ —Se	C [#] —Se	1.970	1.967	0.032	1.948	1.998	21	
Csp ² —Se(2)	C=C—Se—C=C (in tetraselenafulvalene)	1.893	1.895	0.013	1.882	1.902	32	
C _{ar} —Se(3)	Ph ₃ —Se ⁺	1.930	1.929	0.006	1.924	1.936	13	
Csp ³ —Si(5)	C [#] —Si ⁻ —X ₄	1.874	1.876	0.015	1.859	1.884	9	
Csp ³ —Si(4)	CH ₃ —Si—X ₃ C*—Si—X ₃ (C* = CH ₃ excluded) C*—Si—X ₂ (overall)	1.857	1.857	0.018	1.848	1.869	552	
		1.888	1.887	0.023	1.872	1.905	124	
		1.863	1.861	0.024	1.850	1.875	681	
C _{ar} —Si(4)	C _{ar} —Si—X ₃	1.868	1.868	0.014	1.857	1.878	178	
Csp ¹ —Si(4)	C≡C—Si—X ₃	1.837	1.840	0.012	1.824	1.849	8	
Csp ³ —Te	C [#] —Te	2.158	2.159	0.030	2.128	2.177	13	
C _{ar} —Te	C _{ar} —Te	2.116	2.115	0.020	2.104	2.130	72	
Csp ² =Te	see CEDCUJ (2.044)							
Cl—Cl	see PHASCL (2.306, 2.227)							
Cl—I	see CMBIDZ (2.563), HXPASC (2.541, 2.513), METAMM (2.552), BQUINI (2.416, 2.718)							
Cl—N	see BECTAE (1.743–1.757), BOGPOC (1.705)							
Cl—O(1)	in ClO ₄ ⁻	1.414	1.419	0.026	1.403	1.431	252	
Cl—P	(N⋯) ₂ P—Cl (N⋯P aromatic) Cl—P (overall)	1.997	1.994	0.015	1.989	2.004	46	
		2.008	2.001	0.035	1.986	2.028	111	
Cl—S	Cl—S (overall) see also longer bonds in CILSAR (2.283), BIHXIZ (2.357), CANLUY (2.749)	2.072	1.079	0.023	2.047	2.091	6	
Cl—Se	See BIRGUE10, BIRHAL10, CTCNSE (2.234–2.851)							
Cl—Si(4)	Cl—Si—X ₃ (monochloro) Cl ₂ —Si—X ₂ and Cl ₃ —Si—X	2.072	2.075	0.009	2.066	2.078	5	
		2.020	2.012	0.015	2.007	2.036	5	
Cl—Te	Cl—Te in range 2.34–2.60 see also longer bonds in BARRIV, BOJPUL, CETUTE, EPHTEA, OPNTEC10 (2.73–2.94)	2.520	2.515	0.034	2.493	2.537	22	36
F—N(3)	F—N—C ₂ and F ₂ —N—C	1.406	1.404	0.016	1.395	1.416	9	
F—P(6)	in hexafluorophosphate, PF ₆ ⁻	1.579	1.587	0.025	1.563	1.598	72	
P—P(3)	(N⋯) ₂ P—F (N⋯P aromatic)	1.495	1.497	0.016	1.481	1.510	10	

9. BASIC STRUCTURAL FEATURES

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
F—S	43 observations in range 1.409–1.770 in a wide variety of environments F—S(6) in F ₂ —SO ₂ —C ₂ (see FPSULF10, BETJOZ) F—S(4) in F ₂ —S(=O)—N (see BUDTEZ)	1.640 1.527	1.646 1.528	0.011 0.004	1.626 1.524	1.649 1.530	6 24	37
F—Si(6)	in SiF ₆ ²⁻	1.694	1.701	0.013	1.677	1.703	6	
F—Si(5)	F—Si ⁻ —X ₄	1.636	1.639	0.035	1.602	1.657	10	
F—Si(4)	F—Si—X ₃	1.588	1.587	0.014	1.581	1.599	24	
F—Te	see CUCPIZ [F—Te(6) = 1.942, 1.937], FPHTEL [F—Te(4) = 2.006]							
H—N(4)	X ₃ —N ⁺ —H	1.033	1.036	0.022	1.026	1.045	87	21
H—N(3)	X ₂ —N—H	1.009	1.010	0.019	0.997	1.023	95	21
H—O(2)	in alcohols C*—O—H C [#] —O—H in acids O=C—O—H	0.967 0.967 1.015	0.969 0.970 1.017	0.010 0.010 0.017	0.959 0.959 1.001	0.974 0.974 1.031	63 73 16	21 21 21, 38
I—I	in I ₃ ⁻	2.917	2.918	0.011	2.907	2.927	6	
I—N	see BZPRIB, CMBIDZ, HMTITI, HMTNTI, IFORAM, IODMAM (2.042–2.475)							
I—O	X—I—O (see BZPRIB, CAJMAB, IBZDAC11) for IO ₆ ⁻ see BOVMEE (1.829–1.912)	2.144	2.144	0.028	2.127	2.164	6	
I—P(3)	see CEHKAB (2.490–2.493)							†
I—S	see DTHIBR10 (2.687), ISUREA10 (2.629), BZTPPI (3.251)							
I—Te(4)	I—Te—X ₃	2.926	2.928	0.026	2.902	2.944	8	
N(4)—N(3)	X ₃ —N ⁺ —N ⁰ —X ₂ (N ⁰ planar)	1.414	1.414	0.005	1.412	1.418	13	
N(3)—N(3)	(C)(C,H)—N _a —N _b —(C)(C,H) N _a , N _b pyramidal N _a pyramidal, N _b planar N _a , N _b planar overall	1.454 1.420 1.401 1.425	1.452 1.420 1.401 1.425	0.021 0.015 0.018 0.027	1.444 1.407 1.384 1.407	1.457 1.433 1.418 1.443	44 68 40 139	5, 39 40 40 40
N(3)—N(2)	in pyrazole (N1—N2) in pyridazinium (N1 ⁺ —N2)	1.366 1.350	1.366 1.349	0.019 0.010	1.350 1.345	1.375 1.361	20 7	
N(2)≡N(2)	N≡N (aromatic) in pyridazine with C,H as <i>ortho</i> substituents with N,C1 as <i>ortho</i> substituents	1.304 1.368	1.300 1.373	0.019 0.011	1.287 1.362	1.326 1.375	6 9	
N(2)=N(2)	C [#] —N=N—C [#] (<i>cis</i>) (<i>trans</i>) (overall) C _{ar} —N=N—C _{ar} X—N=N=N (azides)	1.245 1.222 1.240 1.255 1.216	1.244 1.222 1.241 1.253 1.226	0.009 0.006 0.012 0.016 0.028	1.239 1.218 1.230 1.247 1.202	1.252 1.227 1.251 1.262 1.237	21 6 27 13 19	
N(2)=N(1)	X—N=N=N (azides)	1.124	1.128	0.015	1.114	1.137	19	
N(3)—O(2)	(C,H) ₂ —N—OH (Nsp ² : planar) C ₂ —N—O—C (Nsp ³ : pyramidal) C ₂ —N—O—C (Nsp ² : planar) in furoxan (N2—O1)	1.396 1.463 1.397 1.438	1.394 1.465 1.394 1.436	0.012 0.012 0.011 0.009	1.390 1.457 1.388 1.430	1.401 1.468 1.409 1.447	28 22 12 14	
N(3)—O(1)	(C≡) ₂ N ⁺ —O ⁻ in pyridine <i>N</i> -oxides in furoxan (⁺ N2—O6 ⁻)	1.304 1.234	1.299 1.234	0.015 0.008	1.291 1.228	1.316 1.240	11 14	

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
N(2)—O(2)	in oximes: (C [#]) ₂ —C=N—OH (H)(Csp ²)—C=N—OH (C [#])(Csp ²)—C=N—OH (Csp ²) ₂ —C=N—OH (C,H) ₂ —C=N—OH (overall)	1.416 1.390 1.402 1.378 1.394	1.418 1.390 1.403 1.377 1.395	0.006 0.011 0.010 0.017 0.018	1.416 1.380 1.393 1.365 1.379	1.420 1.401 1.410 1.393 1.408	7 20 18 16 67	
	in furazan (O1—N2, O1—N5) in furoxan (O1—N5) in isoxazole (O1—N2)	1.385 1.380 1.425	1.383 1.380 1.425	0.013 0.011 0.010	1.378 1.370 1.417	1.392 1.388 1.434	12 14 9	
N(3)=O(1)	in nitrate ions NO ₃ ⁻ in nitro groups: C*—NO ₂ C [#] —NO ₂ C _{ar} —NO ₂ C—NO ₂ (overall)	1.239 1.212 1.210 1.217 1.218	1.240 1.214 1.210 1.218 1.219	0.020 0.012 0.011 0.011 0.013	1.227 1.206 1.203 1.211 1.210	1.251 1.221 1.218 1.215 1.226	105 84 251 1116 1733	
	X ₂ —P(=X)—NX ₂ Nsp ² : planar Nsp ³ : pyramidal (overall)	1.652 1.683 1.662	1.651 1.683 1.662	0.024 0.005 0.029	1.634 1.680 1.639	1.670 1.686 1.682	205 6 358	
	subsets of this group are: O ₂ —P(=S)—NX ₂ C—P(=S)—(NX ₂) ₂ O—P(=S)—(NX ₂) ₂ P(=O)—(NX ₂) ₃	1.628 1.691 1.652 1.663	1.624 1.694 1.654 1.668	0.015 0.018 0.014 0.026	1.615 1.678 1.642 1.640	1.634 1.703 1.664 1.679	9 28 28 78	
	—NX—P(—X)—NX—P(—X)— (P ₂ N ₂ ring) —NX—P(=S)—NX—P(=S)— (P ₂ N ₂ ring) in P-substituted phosphazenes: (N $\ddot{\text{C}}$) ₂ P—N (amino) (aziridiny)	1.730 1.697 1.637 1.672	1.721 1.697 1.638 1.674	0.017 0.015 0.014 0.010	1.716 1.690 1.625 1.665	1.748 1.703 1.651 1.676	20 44 16 15	
	Ph ₃ —P=N ⁺ =P—Ph ₃ Ph ₃ —P=N—C,S	1.571 1.599	1.573 1.597	0.013 0.018	1.563 1.580	1.580 1.615	66 7	
N(2)≡P(3)	N≡P aromatic in phosphazenes in P≡N≡S	1.582 1.604	1.582 1.606	0.019 0.009	1.571 1.594	1.594 1.612	126 36	
	C—SO ₂ —NH ₂ C—SO ₂ —NH—C [#] C—SO ₂ —N—(C [#]) ₂	1.600 1.633 1.642	1.601 1.633 1.641	0.012 0.019 0.024	1.591 1.615 1.623	1.610 1.652 1.659	14 47 38	35 35 35
N(3)—S(2)	C—S—NX ₂ Nsp ² : planar (for Nsp ³ pyramidal see MODIAZ: 1.765) X—S—NX ₂ Nsp ² : planar	1.710 1.707	1.707 1.705	0.019 0.012	1.698 1.699	1.722 1.715	22 30	23 23
	C=N—S—X	1.656	1.663	0.027	1.632	1.677	36	
N(2)≡S(2)	N≡S aromatic in P≡N≡S	1.560	1.558	0.011	1.554	1.563	37	
N(2)=S(2)	N=S in N=S=N and N=S=S	1.541	1.546	0.022	1.521	1.558	37	
N(3)—Se	see COJCUZ (1.830), DSEMOR10 (1.846, 1.852), MORTS10 (1.841)							
N(2)—Se	see SEBZQI (1.805), NAPSEZ10 (1.809, 1.820)							
N(2)=Se	see CISMUM (1.790, 1.791)							
N(3)—Si(5)	see DMESIP01, BOJLER, CASSAQ, CASYOK, CECXEN, CINTEY, CIPBUY, FMESIB, MNPSIL, PNPOSI (1.973–2.344)							
N(3)—Si(4)	X ₃ —Si—NX ₂ (overall) subsets of this group are: X ₃ —Si—NHX	1.748 1.714	1.746 1.719	0.022 0.014	1.735 1.702	1.757 1.727	170 16	

9.5. TYPICAL INTERATOMIC DISTANCES: ORGANIC COMPOUNDS

Table 9.5.1.1. Average lengths (cont.)

Bond	Substructure	<i>d</i>	<i>m</i>	σ	<i>q_l</i>	<i>q_u</i>	<i>n</i>	Note
P(4)—P(4)	$X_3-P-P-X_3$	2.256	2.259	0.025	2.243	2.277	6	
P(4)—P(3)	see CECHEX (2.197), COZPIQ (2.249)							
P(3)—P(3)	$X_2-P-P-X_2$	2.214	2.210	0.022	2.200	2.224	41	
P(4)=P(4)	see BUTSUE (2.054)							
P(3)=P(3)	see BALXOB (2.034)							
P(4)=S(1)	$C_3-P=S$ $(N,O)_2(C)-P=S$ $(N,O)_3-P=S$	1.954 1.922 1.913	1.952 1.924 1.914	0.005 0.014 0.014	1.950 1.913 1.906	1.957 1.927 1.921	13 26 50	
P(4)=Se(1)	$X_3-P=Se$	2.093	2.099	0.019	2.075	2.108	12	
P(3)—Si(4)	$X_2-P-Si-X_3$: 3- and 4-rings excluded (see BOPFER, BOPFIV, CASTOF10, COZVIW: 2.201–2.317)	2.264	2.260	0.019	2.249	2.283	22	
P(4)=Te(1)	see MOPHTE (2.356), TTEBPZ (2.327)							
S(2)—S(2)	$C-S-S-C$ τ (SS) = 75–105° τ (SS) = 0–20° (overall) in polysulfide chain $-S-S-S-$	2.031 2.070 2.048 2.051	2.029 2.068 2.045 2.050	0.015 0.022 0.026 0.022	2.021 2.057 2.028 2.037	2.038 2.077 2.068 2.065	46 28 99 126	
S(2)—S(1)	$X-N=S-S$	1.897	1.896	0.012	1.887	1.908	5	
S—Se(4)	see BUWZUO (2.264, 2.269)							
S—Se(2)	$X-Se-S$ (any)	2.193	2.195	0.015	2.174	2.207	9	
S(2)—Si(4)	$X_3-Si-S-X$	2.145	2.138	0.020	2.130	2.158	19	
S(2)—Te	$X-S-Te$ (any) $X=S-Te$ (any)	2.405 2.682	2.406 2.686	0.022 0.035	2.383 2.673	2.424 2.694	10 28	
Se(2)—Se(2)	$X-Se-Se-X$	2.340	2.340	0.024	2.315	2.361	15	
Se(2)—Te(2)	see BAWFUA, BAWGAH (2.524–2.561)							†
Si(4)—Si(4)	$X_3-Si-Si-X_3$ three-membered rings excluded: see CIHRAM (2.511)	2.359	2.359	0.012	2.349	2.366	42	
Te—Te	see CAHJOK (2.751, 2.704)							

† See opening paragraph of Section 9.5.3. For numbered footnotes, see Appendix 1.